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Using Boyle s law: $P_1V_1 = P_2V_2$, $103 \text{ kPa} \times 5.2 \text{ L} = 400 \text{ kPa} \times V_2$ $V_2 = (103 \text{ kPa} \times 5.2 \text{ L}) / 400 \text{ kPa} = 1.34 \text{ L}$ $P_1 = 700 \text{ mm Hg} \times 1 \text{ atm} / 760 \text{ mm Hg} = 0.921 \text{ atm}$, $V_1 = 200 \text{ ml}$,

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As a mathematical equation, Boyle's Law is written as either: $P_1V_1 = P_2V_2$, or, $P_1/P_2 = V_2/V_1$ and $P \propto 1/V$ is the water pressure at that temperature. Henry's law states that:

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L G Thomson is the author of Boyle's Law (4.67 avg rating, 6 ratings, 4 reviews, published 2014)

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